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# An evaluation of four commercial HPLC chiral detectors: A comparison of three polarimeters and a circular dichroism detector

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#### Abstract

With increasing frequency, new drug candidates being introduced into pharmaceutical drug pipelines are chiral. Often only one enantiomer exhibits the desired biological activity and the other enantiomer may exhibit undesired side effects, thereby making chiral purity an important parameter. The introduction of chiral analysis adds additional complications in drug development. The pharmaceutical industry is constantly striving to streamline processes and improve efficiencies in an effort to move molecules to market quickly.

In order to simplify the process of chiral method development, chiral screening can be set up, however a successful chiral screen depends on optimizing two factors: the column and the detector. The following work investigated the second factor and evaluated two types of commercially available chiral detectors for their possible use in chiral method development and screening: polarimeters and circular dichroism (CD) detectors. Linearity, precision, and the limit of detection (LD) of six compounds (*trans*-stilbene oxide, ethyl chrysanthemate, propranolol, 1-methyl-2-tetralone, naproxen, methyl methionine) on four commercial detectors (three polarimeters and one CD detector) were determined experimentally and the limit of quantitation (LQ) calculated from the experimental LD. *Trans*-stilbene oxide worked well across all the detectors, showing good linearity, precision and low detection limits. However, the other five compounds proved to be more discriminating and showed that the circular dichroism detector performed better as a detector for chiral screens, over the polarimeters.

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#### 1. Introduction

Chiral chromatography is highly dependent on the column, which has seen many recent improvements, and the detector. Chiral detectors have been studied to some extent; however, few studies have done side-by-side comparisons detector responses [1,2]. This study aimed to directly compare the precision, linearity, and limit of detection of two types of commercial chiral detectors, polarimeters and circular dichroism (CD) detectors, to assess their usefulness for chiral screening purposes. Three polarimeters were available to our laboratory for the study, however, there was only one commercially available CD detector. Thorough reviews of CD spectroscopy and how it relates to that of polarimeters is well documented [3], therefore only the concepts as there relate to liquid chromatography detectors are highlighted.

1.1. The detectors: Polarimeters

#### 1.1.1. Polarimeter-1 (PLR-1)

Normal light waves vibrate in many planes; however plane polarized light is generated when normal light is passed through an optical polarizing filter. This effect results in a light beam emerging that vibrates in a single plane (linearly polarized). A compound is optically active if linearly polarized light is rotated when passing through it. The degree of rotation is dependent on both the concentration of a chiral compound and its molecular structure. Every optically active substance has its own specific rotation (degree of rotation in polarized light) as defined by Biots law:

$$[\alpha]_{\lambda}^{T} = \frac{\alpha_{\lambda}^{T}}{cl}$$

where  $[\alpha]$  = specific rotation; l = optical path length in dm;  $\lambda$  = wavelength; T = temperature;  $\alpha$  = optical rotation, and c = concentration in g/mL.

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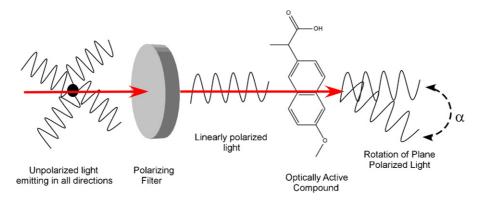


Fig. 1. Optical processes in the polarimeter.

The specific rotation of the molecule, not the absorption characteristics, is what determines the signal strength using the polarimeter. The Polarimeter-1 uses a diode laser at 670 nm as the light source. The vendor has chosen this wavelength as it is in the flat region of the *plain curve* (section of curve where there is little change in  $[\alpha]$ , see Section 1.1.2), a region with little optical interferences. Experimentally, we have found that the polarimeter is the chiral detector most susceptible to scatter, possibly due to use of this long wavelength. Fig. 1 shows a schematic of the optics in the polarimeter.

## 1.1.2. Polarimeter-2 (PLR-2)

The Polarimeter-2 detector is similar in design and function to the Polarimeter-1, with the exception of a light emitting diode (LED) at 426 nm being the light source and having a second polarizing filter present in-line after the sample. The choice of the blue wavelength is based on the *plain curve*, which is a graphic depiction of Drude's equation:

$$[\alpha]_{\lambda} = \frac{\sum A_i}{\lambda_2 - \lambda_i}$$

where  $A_i$  is a molecular constant and  $\lambda_i$  is a constant wavelength [4]. The equation shows the normal behaviour of *optical rotation dispersion* (the dependence of rotational strength of optically active molecules on the wavelength of light used for the measurements) in the absence of chromophores or in spectral regions that are distant from absorption bands. The equation also points out that the angle of rotation, as a function of wavelength, is greatest at shorter wavelengths (see Fig. 2). Therefore, to optimize the chiral response in a molecule, lower source wavelengths yield stronger responses.

However, at lower wavelengths right- and left-handed circularly polarized light<sup>1</sup> propagate at different velocities and are absorbed by molecules to a different extent (this phenomenon is known as circular dichroism; see below). When this happens, it causes a deviation from Drude's equation, known as the "Cotton effect". Fig. 2 shows how the plain optical rotation dispersion curve is affected by the Cotton effect. Fig. 2 also highlights the area where there is maximized specific rotation ( $[\alpha]$ ) and min-

imized Cotton effects, which is in the range of 400–460 nm. Therefore, Polarimeter-2 optimizes these effects by using blue light from a light emitting diode (LED) as the light source.

# 1.1.3. Optical rotary dispersion (ORD) detector

The ORD detector is similar in design and function to the Polarimeter-1; however the light source for this detector is a Xe–Hg lamp, which is readily available and utilizes the strong line emissions of Hg at 365 nm, which can be tuned to cover a spectral range of 350–900 nm, if required.

This detector utilizes the lowest wavelength of the polarimeters (365 nm versus 426 and 670 nm) and therefore one would expect that this detector would give the strongest signal, based on Drude's equation. However, the analog signals collected from these detectors were dependent on the gain set for each detector. Since the gain settings of the three detectors are not comparable, one cannot test Drude's equation with our data set.

## 1.2. The detectors: Circular dichroism (CD)

When an optically active compound preferentially absorbs right or left circularly polarized light, the difference between the right and left absorbances [A(r)-A(1)] (often a very small value) is recorded as the CD signal. As with UV absorbance, the CD signal is wavelength dependent. From the above discussion of the Cotton effect (Section 1.1.2) a molecule should have a chromophore with absorption in the range of 200–420 nm

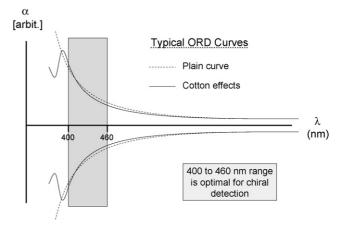


Fig. 2. Graphical depiction of the plain curve and with Cotton effects [5].

<sup>&</sup>lt;sup>1</sup> Circularly polarized light: the two beams of linearly polarized light that are of equal amplitude and are a quarter wave out of phase.

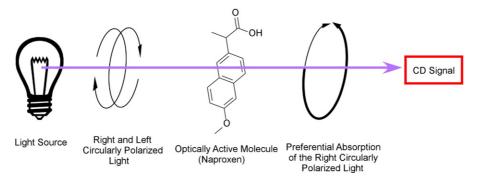


Fig. 3. Optical components of a CD detector.

to have strong CD signal. For the general screening in this study, 230 nm was chosen, as this was the lowest wavelength that would show little interference from the solvents used in the study (UV cut-off for solvents used: *n*-heptane, 220 nm; acetonitrile, 200 nm; water, 190 nm; ethanol, 210 nm; methanol, 220 nm). The wavelength was kept as low as possible as some compounds were specifically chosen to have weak chromophores. It is important to note that 230 nm was not the optimal wavelength for all the samples tested, however it did suffice to yield a signal for each compound. Fig. 3 shows the basic components of the CD detector.

## 2. Experimental

The compounds studied were chosen based on presence or absence of chromophore and whether the compounds require normal or reverse phase chromatographic methods. A Perkin-Elmer high performance liquid chromatography (HPLC) system was used for the assays with the ORD (Jasco Inc., Gurnee,

IL), CD (Jasco Inc., Gurnee, IL) and Polarimeter-2 (IBZ Chiralyzer from Resolution Systems, Holland, MI), which included a Series 200 pump, Series 200 autosampler, a Series 200 peltier column oven and a PE Nelson 600 series LINK box. The chiral detectors were connected to an NCI 900 box and the data was acquired and processed through TotalChrom® operating software. Polarimeter-1 (Advanced Laser Polarimeter, PDR Chiral Inc., Jupiter, FL) was connected to a Shimadzu VP HPLC system including an SCL-10A system controller, FCU-10AL proportioning valves, DGU-14A degassers, an LC-10AD pump, SIL-10AD autosampler, CTO-10AC column oven and an SPD-10A UV-vis detector. This system was controlled by Shimadu Client/Server software.

A 7  $\mu$ L injection of  $\sim$ 1.5 mg/mL stock solution was used for the precision experiments, however linearity was determined by injecting 2–6  $\mu$ L of  $\sim$ 0.5 mg/mL stock solution and 7, 10, 13, 17 and 20  $\mu$ L of  $\sim$ 1.5 mg/mL stock solution. All concentrations are that of the racemic mixture, not the individual enantiomers. The compound structures are compiled in Fig. 4.

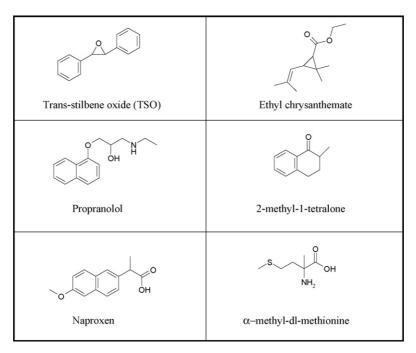


Fig. 4. Structures of compounds used in detector study.

#### 2.1. Trans-stilbene oxide (TSO)

TSO (racemic mix, Fluka Chemie, Switzerland) was a chromophoric compound analyzed by normal phase chiral chromatography. The mobile phase was composed of 90:10 (v/v) heptane (J.T. Baker, Phillipsburg, NJ):ethanol (HPLC grade, Sigma–Aldrich, St. Louis, MO); run isocratically for 20 min, using a Chiralcel OJ column (4.6 mm × 250 mm) at 1 mL/min and a column temperature 27 °C. Samples were prepared in isopropyl alcohol (Sigma).

# 2.2. Ethyl chrysanthemate

Ethyl chrysanthemate (mixture of isomers, Aldrich, St. Louis, MO) is a weak to non-chromophoric compound analyzed by normal phase chiral chromatography. The mobile phase was composed of 100% heptane; run isocratically for 30 min using a Regis Whelko O1 column (4.6 mm  $\times$  250 mm) at 0.5 mL/min and a column temperature of 27 °C. Samples were prepared in isopropyl alcohol (Sigma).

# 2.3. Propranolol

( $\pm$ )Propranolol (Aldrich, St. Louis, MO) was a chromophoric compound analyzed by polar organic chiral chromatography. The mobile phase was composed of 55:45 (v/v) methanol (Sigma–Aldrich):acetonitrile (Sigma–Aldrich) with 0.2% (v/v) triethylamine (J.T. Baker) and 0.3% (v/v) acetic acid (Aldrich); run isocratically for 30 min using a Chirobiotic T column (4.6 mm  $\times$  250 mm) at 1.5 mL/min, and a column temperature of 27 °C. Samples were prepared in isopropyl alcohol (Sigma).

#### 2.4. 2-Methyl-1-tetralone

2-Methyl-1-tetralone (racemic mix, Aldrich) was a chromophoric compound analyzed by reverse phase chiral chromatography. The mobile phase was composed of 60:40 (v/v) water:acetonitrile; run isocratically for 40 min using a Chiralpak AD-RH column (4.6 mm  $\times$  150 mm) at 0.5 mL/min, and a column temperature of 20 °C. Samples were prepared in isopropyl alcohol (Sigma).

## 2.5. Naproxen

Naproxen was a chromophoric compound analyzed by reverse phase chiral chromatography. (R)-Naproxen was received from Toronto Research Chemical (North York, Ont., Canada) and the (S)-naproxen was received from the Caymon Chemical Co. (Ann Arbor, MI). The mobile phase was composed of 75:25 (v/v) water:methanol (Mallinckrodt, Phillipsburg, NJ) with 0.1% (v/v) diethylamine (Fluka) and 0.2% (v/v) acetic acid (Aldrich 99.9%); run isocratically for 40 min using a Chirobiotic R column (4.6 mm  $\times$  250 mm) at 0.8 mL/min, and a column temperature of 27 °C. Samples were prepared in isopropyl alcohol (Sigma).

## 2.6. α-Methyl-DL-methionine

Methyl-DL-methionine (Sigma) was a compound with a weak chromophore analyzed by reverse phase chiral chromatography. The mobile phase was composed of 80:20 (v/v) water:methanol with 0.1% (v/v) diethylamine and 0.2% (v/v) acetic acid (J.T. Baker); run isocratically for 15 min using a Chirobiotic T column (4.6 mm  $\times$  250 mm) at 0.8 mL/min, and a column temperature of 27  $^{\circ}$ C. Samples were prepared in isopropyl alcohol (Sigma).

#### 3. Results and discussion

#### 3.1. Precision

System precision was determined using six consecutive injections of the target concentration of the analyte and the comparing the percent relative standard deviations (%RSD) of the area counts. The target racemate concentration was  $\sim 0.5 \, \text{mg/mL}$ , which was adjusted based upon detector sensitivity. A compilation of the precision data is found in Table 1.

The CD detector yielded the best precision across all compounds studied (a range of  $0.30{\text -}8.46\%\text{RSD}$ ) and for 7 out the 10 experiments, showed a percent relative deviation (%RSD) of less than 5. Ethyl chrysanthemate, propranolol and naproxen at the  $0.5\,\text{mg/mL}$  level showed good precision, which led to the precision being repeated at  ${\sim}10\times$  lower concentrations to better define the capabilities of the detector. TSO was consistently showing better responses with the CD detector, therefore to truly probe the capabilities of the detector, all experiments with TSO and the CD detector used the standard concentration range of  $0.05{\text -}0.15\,\text{mg/mL}$ ,  $10{\text -}\text{fold}$  lower than for the other compounds. For concentrations  ${\sim}0.05$  to  $0.08\,\text{mg/mL}$ , TSO, naproxen and enantiomer 2 of propranolol showed system precision values of less than 5%RSD.

Polarimeter-2 showed a slightly better response over Polarimeter-1 (PLR-2: 0.77–32.39%, PLR-1 range: 1.35–24.99%), with the ORD showing the worst precision data (range: 1.54–43.35%).

The United States Pharmacopeia (USP) outlines system suitability for precision of replicate injections for HPLC as five injections for a %RSD of  $\leq$ 2.0% and six replicates for %RSD of  $\geq$ 2.0% [6]. In general, for most pharmaceutical HPLC assay procedures a %RSD of  $\leq$ 2.0% is the benchmark for chromatographic precision. This being the case, only the CD detector performed as per accepted HPLC detector specifications. In general the polarimeter responses were low at the concentrations studied, which affected the precision %RSD and made them less useful as quantitative HPLC detectors.

# 3.2. Linearity

Linearity was determined by making triplicate injections at each concentration, followed by a regression analysis. The correlation coefficient (r) was used to assess linearity. To better investigate the capabilities of the different types of detectors, linearity was run on what would be considered a standard concentration range expected in a typical drug substance batch

Table 1 Summary of system precision data for the four commercial chiral detectors studied

	TSO	Ethyl chrysanthemate	Propranolol	2-Methyl-1-tetralone	Naproxen	Methyl methionine
Polarimeter-1						
Concentration (mg/mL)	0.5423	0.5018	0.5339	0.5461	0.5296	5.746
%RSD enantiomer 1	1.55	11.22	16.75	18.56	8.26	12.60
%RSD enantiomer 2	1.35	24.99	10.64	9.19	12.98	6.02
Polarimeter-2						
Concentration (mg/mL)	0.5453	0.4914	0.5425	0.4864	0.5184	2.069
%RSD enantiomer 1	0.90	11.29	12.61	8.76	11.39	28.86
%RSD enantiomer 2	0.77	23.72	21.92	5.24	11.11	32.39
ORD						
Concentration (mg/mL)	0.5204	0.4768	0.5223	0.5522	0.5167	2.3002
%RSD enantiomer 1	2.09	7.72	13.54	11.27	31.58	ND
%RSD enantiomer 2	1.54	16.21	17.00	19.89	43.35	ND
CD						
Concentration (mg/mL)	0.0536	0.5104	0.5299	0.6207	0.5241	0.5311
%RSD enantiomer 1	0.30	2.23	1.20	2.99	0.45	8.18
%RSD enantiomer 2	1.03	1.50	2.05	5.10	0.41	8.46
Concentration (mg/mL)		0.0796	0.0825		0.0725	
%RSD enantiomer 1		14.21	13.11		0.85	
%RSD enantiomer 2		5.65	3.77		0.75	

ND = not detected.

(0.5–1.5 mg/mL), as well as on an approximately 10-fold lower concentration set to assess the lower working range of each detector. A 10-fold dilution proved to be an appropriate cut-off for determining the lower linear range for the detectors, as some compounds (ethyl chrysanthemate, naproxen, methyl methionine) could not be detected at this level, while others (TSO) still showed good linearity. Methyl methionine could not be reliably detected by the polarimeters at the 0.5–1.5 mg/mL level; there-

fore the concentration range was increased to range from  $\sim$ 2 to 6 mg/mL, so that linearity and the limits of detection and quantitation could be determined. Table 2 summarizes the linearity data for the higher concentration ranges and Table 3 summarizes the lower concentration linearity data.

Across the standard (higher) concentration range, the responses from ethyl chrysanthemate (enantiomer 1) and naproxen were off-scale at the CD wavelength of 230 nm. Since

Table 2 High concentration linearity data

	TSO	Ethyl chrysanthemate	Propranolol	2-Methyl-1-tetralone	Naproxen	Methyl methionine
Polarimeter-1						
Concentration range (mg/mL)	0.5423-1.5494	0.5249-1.4338	0.5339-1.5254	0.1677-1.5604	0.5296-1.513	2.2984-5.7460
Enantiomer 1 correlation coefficient ( <i>r</i> )	0.9995	Response not linear	0.9801	0.9413	0.9780	0.9409
Enantiomer 2 correlation coefficient ( <i>r</i> )	0.9999	Response not linear	0.9767	0.9527	0.9189	0.9429
Polarimeter-2						
Concentration range (mg/mL)	0.5453-1.5580	0.4914-1.4040	0.1580-1.3175	0.4864-1.3896	0.1527-1.4810	2.0689-5.9110
Enantiomer 1 correlation coefficient ( <i>r</i> )	0.9910	0.8977	0.9738	0.9880	0.9725	0.8650
Enantiomer 2 correlation coefficient ( <i>r</i> )	0.9862	0.9239	0.8692	0.9378	0.9461	0.6373
ORD						
Concentration range (mg/mL)	0.5204-1.4868	0.4768 - 1.3622	0.5223-1.4922	0.5522-1.5777	0.5167-1.4762	2.3002-6.5720
Enantiomer 1 correlation coefficient ( <i>r</i> )	0.9858	0.8567	0.9021	0.9916	0.9346	0.8831
Enantiomer 2 correlation coefficient (r)	0.9887	0.8658	0.9516	0.9662	0.8977	0.8166
CD						
Concentration range (mg/mL)	0.0487-0.1461	0.5104-1.4582	0.5299-1.5140	0.6207-1.7734	0.5241-1.4973	0.5311-1.5174
Enantiomer 1 correlation coefficient ( <i>r</i> )	0.9997	Off-scale at 0.7291 mg/mL	0.9947	0.9847	Off-scale at 0.75 mg/mL	0.9939
Enantiomer 2 correlation coefficient ( <i>r</i> )	0.9888	0.9878	0.9919	0.9912	Off-scale at 0.75 mg/mL	0.9812

Table 3 Low concentration linearity data

	TSO	Ethyl chrysanthemate	Propranolol	2-Methyl-1-tetralone	Naproxen	Methyl methionine
Polarimeter-1						
Concentration range (mg/mL)	0.05378-0.16134	ND	0.12805-0.15366	ND	ND	ND
Enantiomer 1 correlation coefficient ( <i>r</i> )	0.9695	ND	0.9181	ND	ND	ND
Enantiomer 2 correlation coefficient ( <i>r</i> )	0.9806	ND	0.8542	ND	ND	ND
Polarimeter-2						
Concentration range (mg/mL)	0.0578-0.1734	ND	ND	0.1440-0.2160	ND	ND
Enantiomer 1 correlation coefficient $(r)$	0.9825	ND	ND	0.8291	ND	ND
Enantiomer 2 correlation coefficient ( <i>r</i> )	0.9892	ND	ND	0.7900	ND	ND
ORD						
Concentration range (mg/mL)	0.04376-0.13128	ND	ND	0.0543-0.1629	ND	ND
Enantiomer 1 correlation coefficient ( <i>r</i> )	0.9382	ND	ND	0.9138	ND	ND
Enantiomer 2 correlation coefficient ( <i>r</i> )	0.9207	ND	ND	0.8685	ND	ND
CD						
Concentration range (mg/mL)	0.00487-0.01461	0.06274-0.18822	0.05108-0.15324	0.02358-0.23584	0.02207-0.06621	0.04914-0.14742
Enantiomer 1 correlation coefficient ( <i>r</i> )	0.9992	0.9893	0.9946	0.9834	0.9990	0.8897
Enantiomer 2 correlation coefficient (r)	0.9967	0.9907	0.9959	0.9919	0.9932	ND

ND = not detected.

the wavelength was not optimized for any of the study compounds, the linear response may be further improved upon optimizing the choice of wavelength. As stated earlier, the intended final use for the chiral detectors was for general screening and 230 nm was chosen as a constant wavelength for this investiga-

tion, as this was the lowest wavelength that would show little interference from the solvents used (i.e. more compounds are UV active at lower wavelengths). It is evident that 230 nm is not optimal for naproxen and propranolol as both enantiomers show a positive response, nor for 2-methyl-1-tetralone as both

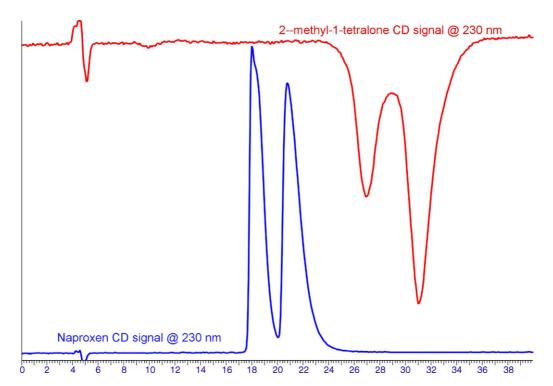


Fig. 5. CD chromatograms of naproxen and 2-methyl-1-tetralone at 230 nm showing that the signal is not optimized.

enantiomers show a negative response (Fig. 5). The responses of these compounds at 230 nm are only a snapshot of the whole CD spectrum. The predominant use of CD spectroscopy to date gave general information about the environment of the chiral center, and the determination of secondary structure in larger molecules, such as the location of ketone groups, the degree of coiling in protein helices and the type of substitution in amino acids [7]. Many factors can contribute to the observed responses in Fig. 5 but when a CD wavelength is optimized, enantiomers yield one positive and one negative signal. Since the scope of this study was to test the feasibility of the detectors for general screening, further investigations into the CD responses were not pursued. For the compounds that did remain on-scale (TSO, propranolol, 2-methyl-1-tetralone and methyl methionine), the CD detector showed the most linear response overall (range of r: 0.9847-0.9997).

Polarimeter-1 showed the next best linear response (range: r=0.9189–0.9999), followed by Polarimeter-2 (range: r=0.6373–0.9910) and ORD (range: r=0.8166–0.9887) which showed approximately the same linear responses across the compounds. The correlation coefficients determined for the polarimeter detectors were comparably lower than the CD detector slipping into the 0.8 range for the non-chromophoric compounds and even as low as 0.64 in the case of methyl methionine. The response of the polarimeters was lesser (i.e. much lower peak heights) and noisier, than the CD detector responses, making integration more difficult and less consistent between data sets.

The lower concentration range data highlighted the greater sensitivity of the CD detector, which yielded linear responses for all of the compounds investigated. All the polarimeters showed linear responses for TSO.

# 3.3. Limit of detection (LD)

The third method of assessment of the four chiral detectors was the determination of the limit of detection. This was done both experimentally and mathematically. The LD was determined mathematically by the formula [8]:

$$LD = \frac{3.3s}{S}$$

where s = standard deviation of the sample repeats and S = the slope of the calibration curve. When it was possible the calculated LD was tested by injecting samples at the calculated concentration, to test the reliability of the calculated value. The LD data is summarized in Table 4.

Across the entire compound set, the CD detector showed the lowest measured detection limits, ranging from 4.65 µg/mL to 0.098 mg/mL. The limits of detection for the non-chromophoric compounds ethyl chrysanthemate and methyl methionine were on the high end of the range (0.23 mg/mL for ethyl chrysanthemate enantiomers and 0.049 and 0.098 mg/mL for methyl methionine), however they were still lower than those determined by the polarimeter detectors, whose responses are independent of presence of chromophores in the molecule. For most compounds the calculated LD was the same order of magnitude as the observed. For TSO and naproxen, the calculated values were lower than any concentration tested, however the signals were still strong at the lowest concentrations injected, thereby suggesting that the calculated LD values may be realistic. Polarimeter-2 showed the next best limit of detection, ranging from 0.058 to 2.069 mg/mL. Because of its large signal responses, Polarimeter-2 tended to show more scatter at the lower concentrations. Due to this, Polarimeter-2 is considered

Table 4
Limits of detection for the six study compounds on the four chiral detectors

	TSO	Ethyl chrysanthemate	Propranolol	2-Methyl-1-tetralone	Naproxen	Methyl methionine
Polarimeter-1						
Enantiomer 1 LD calculated	6.72 µg/mL	0.978	0.147	0.363	0.166	1.056
Enantiomer 1 LD measured	0.053	0.502	0.154	0.546	0.126	2.298
Enantiomer 2 LD calculated	0.031	0.670	0.082	0.285	0.397	0.361
Enantiomer 2 LD measured	0.053	0.502	0.154	0.546	0.126	2.298
Polarimeter-2						
Enantiomer 1 LD calculated	1.07 μg/mL	0.355	0.201	0.116	0.245	3.368
Enantiomer 1 LD measured	0.058	0.491	0.543	0.144	0.153	2.069
Enantiomer 2 LD calculated	0.015	0.231	0.298	0.150	0.259	4.773
Enantiomer 2 LD measured	0.058	0.491	0.543	0.144	0.153	2.069
ORD						
Enantiomer 1 LD calculated	0.059	0.610	0.328	0.049	0.267	2.053
Enantiomer 1 LD measured	0.044	0.477	0.522	0.054	0.517	5.586
Enantiomer 2 LD calculated	0.071	0.660	0.137	0.088	0.525	1.695
Enantiomer 2 LD measured	0.044	0.477	0.522	0.054	0.517	5.586
CD						
Enantiomer 1 LD calculated	$0.13 \mu g/mL$	11.80 μg/mL	5.26 µg/mL	0.021	$2.76 \mu g/mL$	0.028
Enantiomer 1 LD measured	4.87 μg/mL	0.023	4.65 μg/mL	0.024	0.022	0.049
Enantiomer 2 LD calculated	0.27 μg/mL	0.027	8.95 μg/mL	0.019	4.29 μg/mL	0.421
Enantiomer 2 LD measured	4.87 μg/mL	0.023	4.65 μg/mL	0.024	0.022	0.098

Units are in mg/mL unless otherwise noted.

Table 5
Limits of quantitation for the six study compounds on the four chiral detectors

	TSO	Ethyl chrysanthemate	Propranolol	2-Methyl-1-tetralone	Naproxen	Methyl methionine
Polarimeter-1						
Enantiomer 1 LQ calculated	0.022	3.259	0.490	1.209	0.553	3.519
Enantiomer 2 LQ calculated	0.103	2.232	0.273	0.948	1.324	2.102
Polarimeter-2						
Enantiomer 1 LQ calculated	3.57 µg/mL	1.183	0.670	0.387	0.816	11.227
Enantiomer 2 LQ calculated	0.048	0.768	0.994	<b>0</b> .500	0.864	15.910
ORD						
Enantiomer 1 LQ calculated	0.195	2.035	1.095	0.163	0.892	6.845
Enantiomer 2 LQ calculated	0.236	2.201	0.457	0.295	1.749	5.649
CD						
Enantiomer 1 LQ calculated	$0.45 \mu g/mL$	0.039	0.018	0.069	9.20 μg/mL	0.094
Enantiomer 2 LQ calculated	0.89 μg/mL	0.089	0.030	0.065	14.30 μg/mL	1.405

Units are in mg/mL unless otherwise noted.

a sensitive detector but not precise. The lack of precision, however, does not deter its use of this detector for screening purposes. Polarimeter-1 and the ORD ranked third and fourth in sensitivity, respectively (PLR-1 range: 0.053–2.298 mg/mL; ORD range: 0.044–5.586 mg/mL).

# 3.4. Limit of quantitation (LQ)

The LQ can also be determined mathematically from the previously determined LD values. The International Conference on Harmonization (ICH) guidelines define LD based on signal to noise (S/N) as  $3 \times S/N$  and LQ as  $10 \times S/N$  [8]. Therefore

$$LD = \frac{3.3s}{S} = 3\frac{S}{N}$$

and LO is

$$LQ = 10\frac{S}{N} = 3.3 LD$$

Using this relationship, the LQ values in Table 5 were calculated from the calculated LD values in Table 4.

Table 5 shows that the standard concentration range used (0.5–1.5 mg/mL) was appropriate for TSO, most 2-methyl-1-tetralone analyses and all CD analyses (highlighted with bold-face type), with the exception of methyl methionine, as the standard concentration range is close to the calculated LQ's for one of the enantiomers. Methyl methionine and other individual results that have calculated LQ's within the standard concentration range are italicized. These data correlate well with the low concentration linearity data (~0.05 to 0.15 mg/mL) in Table 3, as most of the bold marked analyses in Table 5 resulted in detectable linearity data across the lower concentration range.

# 3.5. Achiral use

Use of chiral detection for achiral liquid chromatographic separations has been reported in the literature [9]. Initial studies using these detectors in our laboratory did not yield adequate

sensitivity for pharmaceutical enantiomeric purity determinations. Investigations in this chiral detector application are on going.

#### 4. Conclusion

Two types of chiral detectors were tested and compared to determine their usefulness for chiral method development and screening; polarimeters and CD detectors. Of greatest initial importance, especially for screening work, is the limit of detection, for if a racemic mixture of enantiomers is not available, screening may have to be done on drug substances with only a small amount of the chiral impurity. The CD detector showed the lowest detection limits across all the compounds and modes of chromatography (normal, reverse, polar organic). Since CD detection depends on the difference in absorption of circularly polarized light, and polarimeters are less reliant on the presence of chromophores, compounds with weak chromophores were included in the study, with the intent to truly test the universality of the detectors. Contrary to what had been expected, the CD detector worked well for weakly chromophoric compounds and resulted in similar detection limits as compounds with strong chromophores. Of the polarimeters tested, Polarimeter-2 showed the best detection limits, although there was much scatter at the lower concentrations.

For the precision experiments, the CD detector showed the lowest percent relative standard deviation values at the target concentration of  $\sim\!0.5\,\text{mg/mL}$  and was consistently precise for TSO, ethyl chrysanthemate, propranolol, and naproxen at  $\sim\!0.05\,\text{mg/mL}$ . Of the polarimeters, Polarimeter-2 showed slightly better precision over Polarimeter-1, with the ORD detector showing the worst precision.

The CD detector also showed the best linearity for all compounds, albeit the wavelength used was not optimized for each analyte. The sensitivity of this detector is underscored by the signal; in some cases being so strong at the higher concentration range that the peaks were off-scale and that the data was linear even after a 10-fold dilution of the standards.

Gradients worked well on the CD, Polarimeter-1 and ORD polarimeters; however, Polarimeter-2 had more trouble with gradients (data not shown). To be able to use a gradient, the detector needs to be zeroed at the mid-point of the gradient, followed by adjusting the sensitivity such that the early and late portions of the gradient remained on scale. This makes Polarimeter-2 a little more cumbersome for gradient screens, unless the same gradient is used with many different columns. Blank runs can be programmed in a sequence for changing the gradient and re-zeroing the system. As with a refractive index detector, Polarimeter-2 is best used with isocratic elution runs.

Polarimeter-2 had the best control of all of the detectors over signal gain. When the detector is set at the most sensitive setting, the peak area counts are  $\sim 100$ -fold larger than the other three detectors, making the noise higher, but also making it easier to discern and quantify peaks.

Overall the CD detector has shown to be a general but sensitive chiral detector for chiral method development and screening. In most cases once chiral separation has been established methods are developed using more conventional detection techniques (i.e. UV), however the CD which is equipped with UV detection as well, could possibly be used more universally for chiral analyses as its response is both linear and sensitive. Of the polarimeters the Polarimeter-2 and Polarimeter-1 performed about the same, but came in a distant second to the CD. Polarimeter-2 could prove more useful if the control of the instrument was better interfaced with a computer for controlling

the sensitivity and zeroing the instrument on changing mobile phases.

Based on the overall performance of the four detectors, four detectors are ranked in the following order: CD; Polarimeter-2/Polarimeter-1; ORD.

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## References

- D.R. Bobbit, S.W. Linder, Trends Anal. Chem. 20 (2001) 111–123 (and review articles cited within).
- [2] C. Roussel, N. Vanthuyne, M. Serradeil-Albalat, J.-C. Vallejos, J. Chromatogr. A 995 (2003) 79–85.
- [3] N. Berova, K. Nakanishi, R.W. Woody (Eds.), Circular Dichroism: Principles and Applications, 2nd ed., Wiley-VCH, New York, 2000, pp. 28–35, 820–827
- [4] G. Snatzke, Angew. Chem. Int. Ed. 7 (1968) 14-25 (in English).
- [5] IBZ Messtechnik website: http://www.ibzmesstechnik.de/news.
- [6] USP 28, United States Pharmacopeial Convention Inc., Philladelphia, PA, 2005, Chromatography (Chapter 621).
- [7] D.A. Skoog, Principles of Instrumental Analysis, 3rd ed., Saunders College Publishing, USA, 1985, pp. 399–404.
- [8] Validation of Analytical Procedures: Methodology Q2B (1996) from the International Conference on Harmonization of Technical Requirements for the Registration of Pharmaceuticals for Human Use, 1996.
- [9] M.R. Hadley, G.D. Jonas, Enantiomer 5 (2000) 357-368.